

Thermal Physics, Autumn 2019 CMI

Problem set 3

Due by the beginning of lecture on Tuesday, Sep 10, 2019

First law, Adiabatic model for atmospheric temperature and pressure profile

1. ⟨5⟩ Use the first law to show that the difference in heat capacities is given by

$$C_p - C_V = \left(\left(\frac{\partial U}{\partial V} \right)_T + p \right) \left(\frac{\partial V}{\partial T} \right)_p. \quad (1)$$

2. ⟨2⟩ Now suppose a gas satisfies the ideal equation of state $pV = nRT$ and the ‘Caloric condition’. Use the above to evaluate the difference $C_p - C_V$ for such an ideal gas.
3. ⟨20⟩ It is known that temperature, density and pressure all decrease with height z in the atmosphere. We wish to find the temperature and pressure gradients in a simple model for the atmosphere. We assume that the air is an ideal gas satisfying the ideal gas law $pV = nRT$. Since air is a poor conductor we will assume that parcels of air move without much heat exchange with surroundings (i.e. adiabatically), so we may assume that any two among p, V, T satisfy the adiabatic relation. We will assume a steady state where layers of air in the atmosphere are in equilibrium due to a balance of their weight by the upward pressure gradient.
- (a) ⟨3⟩ Find the differential condition for mechanical equilibrium of a thin layer of air of surface area A at height z and thickness dz . Denote the pressure and density by $p(z)$ and $\rho(z)$.
- (b) ⟨3⟩ Show that the variation of pressure satisfies the equation (μ is average molar mass for air and g the acceleration due to gravity)

$$\frac{dp}{p} = -\frac{\mu g}{RT} dz \quad (2)$$

- (c) ⟨5⟩ Show that the temperature gradient dT/dz is a constant κ (assuming g is constant). Find an analytic formula for κ in terms of the adiabatic index γ .
- (d) ⟨3⟩ Find the numerical value of the temperature gradient by taking reasonable values for the constants.
- (e) ⟨3⟩ Find a differential equation for the variation of pressure with height in the form $dp/dz = f(p, z)$, assuming temperature on the Earth’s surface ($z = 0$) is T_0 .
- (f) ⟨3⟩ Find a formula for $p(z)$, assuming the pressure at $z = 0$ is p_0